SUMMARY OF PRODUCT CHARACTERISTICS

1 NAME OF THE MEDICINAL PRODUCT
Medical Oxygen 100% Medicinal gas, compressed

2 QUALITATIVE AND QUANTITATIVE COMPOSITION
Oxygen (O₂) 100% v/v
(150, 200 bar or 300 bar, 15°C)
For the full list of excipients, see section 6.1.

3 PHARMACEUTICAL FORM
Medicinal gas, compressed.
Oxygen is a colourless, odourless and tasteless gas.
In liquid state it has a blue colour.

4 CLINICAL PARTICULARS
4.1 Therapeutic indications
Normobaric oxygen therapy:
- Treatment or prevention of acute or chronic hypoxia.
- Treatment of cluster headache.

Hyperbaric oxygen therapy
- Treatment of serious carbon monoxide poisoning. (In the case of carbon monoxide poisoning, hyperbaric oxygen therapy is considered essential for patients who have lost consciousness; neurological symptoms, cardiovascular failure or serious acidosis; or pregnant patients (all of these indications irrespective of COHb content)).
- Treatment of decompression sickness, or of air/gas embolism of a different origin.
- As supporting treatment in cases of osteoradionecrosis.
- As supporting treatment in cases of clostridial myonecrosis (gas gangrene).

4.2 Posology and method of administration
Posology
The concentration, flow and duration of the treatment will be determined by a physician, according to the characteristics of each pathology.
Hypoxemia refers to a condition where the arterial partial pressure of oxygen (PaO₂) is lower than 10 kPa (<70 mmHg). An oxygen pressure level of 8 kPa (55 / 60 mmHg) will result in respiratory insufficiency.
Hypoxemia is treated by enriching the patient’s inhalation air with extra oxygen. The
decision to introduce oxygen therapy depends on the degree of hypoxemia and the patient's individual tolerance level.

In all cases, the objective of the oxygen therapy is to maintain a PaO$_2$ > 60 mm Hg (7.96 kPa) or oxygen saturation in the arterial blood ≥ 90%.

If oxygen is administered diluted in another gas, the oxygen concentration in the inspired air (FiO$_2$) must be at least 21%.

**Oxygen therapy at normal pressure (Normobaric oxygen therapy):**

Administration of oxygen should be performed cautiously. The dose should be adapted to the individual needs of the patient, oxygen tension should remain higher than 8.0 kPa (or 60 mmHg) and oxygen saturation of haemoglobin should be > 90%. Regular monitoring of arterial oxygen tension (PaO$_2$) or pulsoxymetry (arterial oxygen saturation (SpO$_2$)) and clinical signs is necessary. The aim is always to use the lowest possible effective oxygen concentration in the inhaled air for the individual patient, which is the lowest dose to maintain a pressure of 8 kPa (60 mmHg)/saturation > 90%. Higher concentrations should be administered as short as possible accompanied by close monitoring of blood gas values.

Oxygen can be administered safely in the following concentrations, for the periods indicated:

- **Up to 100%** less than 6 hours
- **60-70%** 24 hours
- **40-50%** during the second 24-hour period

Oxygen is potentially toxic after two days in concentrations in excess of 40%.

Neonates are excluded from these guidelines because retrolental fibroplasia occurs with a much lower FiO$_2$. The lowest effective concentrations should be sought in order to achieve an adequate oxygenation appropriate for neonates.

- **Spontaneously breathing patients:**
  The effective oxygen concentration is at least 24%. Normally, a minimum of 30% oxygen is administrated to ensure therapeutic concentrations with a safety margin.

  The therapy with high oxygen concentration (> 60%) is indicated for short periods in case of serious asthmatic crisis, pulmonary thromboembolism, pneumonia and alveolitic fibrosis, etc.

  A low oxygen concentration is indicated for the treatment of patients with chronic respiratory insufficiency due to a chronic obstructive upheaval of the airways or other causes. The oxygen concentration must not be more than 28%, for some patients even 24% can be excessive.

  Administration of higher oxygen concentrations (in some cases up to 100%) is possible, although when using most administration devices it is very difficult to obtain concentrations > 60% (80% in the case of children).

  The dose should be adapted to the individual needs of the patient, at flow rates ranging from 1 to 10 litres of gas per minute.

- **Patients with chronic respiratory insufficiency:**
  Oxygen must be administered at flow rates ranging from 0.5 to 2 liters/minute, rates should be adjusted on the basis of blood gas values. The effective oxygen
concentration will be kept below 28% and sometimes even lower than 24% in patients suffering from breathing disorders who depend on hypoxia as a breathing stimulus.

- **Chronic respiratory insufficiency resulting from Chronic Obstructive Pulmonary Disease (C.O.P.D.) or other conditions:**
The treatment is adjusted on the basis of blood gas values. Arterial partial oxygen pressure (PaO₂) should be > 60 mm Hg (7.96 kPa) and oxygen saturation in the arterial blood ≥ 90%.

The most common administration rate is 1 to 3 liters/minute for 15 to 24 hours/day, also covering paradoxical sleep (the most hypoxemia-sensitive period within a day). During a stable disease period, CO₂ concentrations should monitored twice every 3-4 weeks or 3 times per month as CO₂ concentrations can increase during oxygen administration (hypercapnia).

- **Patients with acute respiratory insufficiency:**
Oxygen must be administered at a rate ranging from 0.5 to 15 liters/minute, flow rates should be adjusted on the basis of blood gas values. In case of emergency, considerably higher doses (up to 60 liters/minute) are required in patients with severe respiratory difficulties.

- **Mechanically ventilated patients:**
If oxygen is mixed with other gases, the oxygen fraction in the inhaled gas mixture (FiO₂) may not fall under 21%. In practice, 30% tends to be used as the lower limit. If necessary, the inhaled oxygen fraction can be raised to 100%.

- **Paediatric population:**
New-born infant:
In new-born infant, concentrations of up to 100% can be administered in exceptional cases; however, the treatment must be closely monitored. The lowest effective concentrations should be sought in order to achieve an adequate oxygenation. As a rule, oxygen concentrations in excess of 40% in inhalation air must be avoided, considering the risk of eye damage (retinopathy) or pulmonary collapse. Oxygen pressure in the arterial blood must be closely monitored and kept below 13.3 kPa (100 mmHg). Fluctuations in oxygen saturation should be avoided. By preventing substantial fluctuations in oxygenation, the risk of eye damage can be reduced. (Also see section 4.4.)

- **Cluster headache:**
In the case of cluster headache, 100% oxygen is administered at a flow rate of 7 liters/minute for 15 minutes using a close-fitting facial mask. The treatment should begin in the earliest stage of a crisis.

Hyperbaric oxygen therapy:
**Dosage and pressure should always be adapted to the patient’s clinical condition and therapy should only be given after doctor’s advice. However, some recommendations based on current knowledge are given below.**
Hyperbaric oxygen therapy is done at pressures higher than 1 atmosphere
(1.013 bars) between 1.4 and 3.0 atmosphere (usually anywhere between 2 and 3 atmosphere). Hyperbaric oxygen is administered in a special pressure room. Oxygen therapy at high pressure can also be given using a close-fitting facial mask with a hood covering the head, or through a tracheal tube. Each treatment session lasts 45 to 300 minutes, depending on the indication. Acute hyperbaric oxygen therapy may sometimes last just one or two sessions, whereas chronic therapy may take up to 30 or more sessions. If necessary, the sessions can be repeated two to three times a day.

- Carbon monoxide poisoning:
  Oxygen should be given in high concentrations (100%) as soon as possible following carbon monoxide poisoning until the carboxyhaemoglobin concentration has fallen below dangerous levels (around 5%). Hyperbaric oxygen (starting at 3 atmospheres) is indicated for patients with acute CO poisoning or have exposure intervals ≥24 hours. In addition, pregnant patients, patients with loss of consciousness or higher carboxyhemoglobin levels warrant hyperbaric oxygen therapy. Normobaric oxygen should not be used between multiple hyperbaric oxygen treatments as this can contribute to toxicity. Hyperbaric oxygen seems to also have potential in the delayed treatment of CO poisoning using multiple treatments of low dose of oxygen.

- Patients with decompression sickness:
  Rapid treatment at 2.8 atmosphere is recommended, repeated up to ten times if symptoms persist.

- Patients with air embolism:
  In this case, the dosage is adapted to the patient’s clinical condition and blood gas values. The target values are: PaO₂ > 8 kPa, or 60 mmHg, haemoglobin saturation > 90%.

- Patients with osteoradionecrosis:
  Hyperbaric oxygen therapy in radiation injury usually consist of daily 90-120 min sessions at 2.0-2.5 atmosphere for about 40 days.

- Patients with clostridial myonecrosis:
  It is recommended that a 90-min treatment should be given at 3.0 atmosphere in the first 24h, followed by twice-daily treatments for 4-5 days, until clinical improvement is seen.

**Method of administration**

**Normobaric oxygen therapy**

Oxygen is administered through inhaled air, preferably using dedicated equipment (e.g., a nose catheter or facial mask) via this equipment, oxygen is administered with inhaled air. The gas plus any excess oxygen subsequently leaves the patient in the exhaled air, and mixes with the ambient air (“non-rebreathing” system). In many cases, during anaesthesia special systems with a rebreathing system or recycling system are used so that the exhaled air is inhaled once again (“rebreathing” system).

If the patient cannot breathe independently, artificial breathing support can be provided.

In addition, oxygen can be injected into the bloodstream directly using a so-called
The application of extracorporeal gas exchange devices facilitate oxygenation and decarboxylation without the harm associated with aggressive mechanical ventilation strategies. The oxygenator, which acts as an artificial lung, provides improved oxygen transfer and therefore, blood gas levels are kept within clinical acceptable ranges. After recovery of lung function extracorporeal blood and gas flow is reduced and eventually, stopped. This happens, for example, during cardiac surgery using a cardio-pulmonary by-pass system, as well as in other circumstances that require extracorporeal circulation including acute respiratory insufficiency.

**Hyperbaric oxygen therapy**

Hyperbaric oxygen therapy is administered in a specially constructed pressure room where the ambient pressure can be increased to up to three times the atmospheric pressure. Hyperbaric oxygen therapy can also be provided through a close-fitting facial mask with a hood covering the head, or through a tracheal tube.

### 4.3 Contraindications

**Normobaric oxygen therapy**

There are no absolute contraindications for normobaric oxygen therapy.

**Hyperbaric oxygen therapy**

One absolute contraindication for hyperbaric oxygen therapy is an untreated pneumothorax, including restrictively treated pneumothorax (without a chest tube).

### 4.4 Special warnings and precautions for use

Low oxygen concentrations must be used for patients with respiratory failure who depend on hypoxia as a breathing incentive. In these cases, careful monitoring of the treatment is required, by measuring the arterial oxygen tension (PaO₂) or through pulsoxymetry (arterial oxygen saturation (SpO₂)) and clinical assessment. Special caution is required in the treatment of new-born infant and pre-term new-born infant. In these cases, the lowest effective concentration must be used in order to achieve an adequate oxygenation appropriate for neonates and fluctuations in oxygen saturation should be avoided. Such caution is to minimise the risk of eye damage, retrolental fibroplasia or other potential adverse events, but still while achieving an adequate oxygenation appropriate for neonates and avoiding fluctuations in oxygen saturation.

Arterial oxygen pressure must be closely monitored and should be kept below 13.3 kPa (100mmHg).

High oxygen concentrations in the inhaled air or gas will cause the concentration and pressure of nitrogen to fall. This will also reduce the concentration of nitrogen in tissues and the lungs (alveoli). If oxygen is absorbed into the blood through the alveoli faster than it is supplied through ventilation, the alveoli may collapse.
(atelectasis). This may obstruct the oxygenation of the arterial blood, because no gases are exchanged despite perfusion.

In patients with reduced sensitivity to carbon dioxide pressure in arterial blood, high oxygen levels may cause carbon dioxide retention. In extreme cases, this may lead to carbon dioxide narcosis.

Hyperbaric oxygen therapy must be administered by nursing staff who are qualified for that purpose. Compression and decompression treatment must be carefully phased to minimise the risk of pressure-induced injury (barotrauma).

Preferably, hyperbaric oxygen therapy should not be used for patients with:

- COPD or pulmonary emphysema
- infections of the upper respiratory tract
- recent middle ear surgery
- recent thoracic surgery
- uncontrolled high fever
- serious epilepsy

Caution should be exercised in patients with claustrophobia.

In addition, caution is called for in patients with a medical history of thoracic surgery or epileptic fits.

Caution should be exercised in patients with pre-existing cardiac disorders (mainly heart failure and ischemic heart disease, whether acute – myocardinfarct, or chronic – post revascularization) in case of hyperoxia, as this could potentially lead to worse outcomes.

In patients presenting with a pneumothorax treated with a chest tube and/or patients with a medical history of pneumothorax, the use should be evaluated for each individual patient with regard to the risk on a new (tension) pneumothorax. The treatment with hyperbaric oxygen in patients with a pneumothorax treated with a chest tube should be done in a situation where supportive care can be provided immediately such as in a hospital setting. Caution is called for in patients with a history of thoracic surgery or epileptic fits.

The pulmonary toxicity associated with drugs such as bleomycin, amiodarone, furadantin and similar antibiotics may be exacerbated by inhalation of increased concentration of oxygen. (See also section 4.5 of this document.)

Whenever oxygen is used, the increased risk for spontaneous ignition should be taken into account. This risk is increased in procedures involving diathermy, defibrillation/electro conversion therapy.

### 4.5 Interaction with other medicinal products and other forms of interaction

The pulmonary toxicity associated with drugs such as bleomycin, amiodarone, furadantin and similar antibiotics, may be exacerbated by inhalation of increased concentration of oxygen.

There are reports of interaction with amiodarone. Relapse of pulmonary damage induced by bleomycin or actinomycin may be fatal.
In patients who have been treated for oxygen radical-induced pulmonary damage, oxygen therapy may exacerbate that damage, for example in the treatment of paraquat poisoning.

Oxygen may also aggravate alcohol-induced respiratory depression.

Medicinal products known to provoke adverse events include: adriamycin, menadion, promazine, chlorpromazine, thioridazine and chloroquine. The effects will be particularly pronounced in tissues with high oxygen levels, especially the lungs.

Corticosteroids, sympathicomimetics or X-rays may increase the toxicity of oxygen. Hyperthyroidism or a lack of vitamin C, vitamin E or glutathione may also produce that effect.

4.6 Fertility, pregnancy and lactation

Women that can be pregnant

In case pregnancy cannot be excluded, hyperbaric oxygen should only be used if strictly necessary (for further information see “pregnancy”)

Pregnancy

A limited amount of data from documented experience of the use of (hyperbaric) oxygen therapy in pregnant women indicate no malformative or feto/neonatal toxicity. The available clinical data is insufficient to exclude a risk. Studies in animals, have shown reproductive toxicity after administration of oxygen at increased pressure and in high concentrations (see section 5.3). Low concentrations of normobaric oxygen can be administered safely during pregnancy, if necessary. The use of high concentrations of oxygen and hyperbaric oxygen may be considered in the case of vital indications during pregnancy.

Hyperbaric oxygen should only be used in pregnancy if strictly necessary due to a potential risk of oxidative stress-induced damage in the foetus. In severe carbon monoxide intoxication the benefit vs. risk seems reassuring for the use of hyperbaric oxygen. The use should then be evaluated for each individual patient.

Lactation

Medicinal oxygen can be used during lactation without risks to the infant.

Fertility

There are no data available regarding potential effects of oxygen treatment on male or female fertility.

4.7 Effects on ability to drive and use machines

Oxygen has no influence on the ability to drive and use machines.
## 4.8 Undesirable effects

<table>
<thead>
<tr>
<th>Nervous system disorders</th>
<th>Uncommon (≥1/1000 to &lt;1/100)</th>
<th>Rare (≥1/10,000 to &lt;1/1000)</th>
<th>Very rare (&lt;1/10,000)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Hybaric oxygen treatment</td>
<td>Anxiety; Confusion; Loss of consciousness; Epilepsy unspecified; Toxicity of the central nervous system including nausea, dizziness, spasms, pulmonary toxicity and reversible visual changes</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### Eye disorders

| Eye disorders | Retrolental fibroplasia in neonates who have been exposed to high oxygen concentrations |

<table>
<thead>
<tr>
<th>Ear and labyrinth disorders</th>
<th>Hyperbaric oxygen therapy</th>
</tr>
</thead>
<tbody>
<tr>
<td>Feeling of pressure in the middle ear; Tympanic membrane rupture</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Respiratory, thoracic and mediastinal disorders</th>
<th>Atelectasis; Pleuritis</th>
</tr>
</thead>
<tbody>
<tr>
<td>Respiratory distress syndrome</td>
<td></td>
</tr>
</tbody>
</table>

### Reporting of suspected adverse reactions

Reporting suspected adverse reactions after authorisation of the medicinal product is important. It allows continued monitoring of the benefit/risk balance of the medicinal product. Healthcare professionals are asked to report any suspected adverse reactions via the Yellow Card Scheme

(Website: [www.mhra.gov.uk/yellowcard](http://www.mhra.gov.uk/yellowcard))

## 4.9 Overdose

The toxic effects of oxygen vary according to the pressure of the inhaled oxygen and the duration of exposure. Low pressure (0.5 to 2.0 bar) is more likely to cause pulmonary toxicity than toxicity to the central nervous system. The opposite applies to higher pressure levels (hyperbaric oxygen therapy).
The symptoms of pulmonary toxicity include hypoventilation, coughing and chest pain.

The symptoms of central nervous system toxicity include nausea, dizziness, anxiety and confusion, muscle cramp, loss of consciousness and epileptic fits.

Cases of overdose must be treated by reducing the concentration of inhaled oxygen. In addition, therapy must be provided to maintain the patient’s normal physiological functions (such as breathing support in the case of respiratory depression).

5 PHARMACOLOGICAL PROPERTIES

5.1 Pharmacodynamic properties
Pharmacotherapeutic group: Medical gases, ATC code V03AN01

Oxygen is vital to living organisms, and all tissues must be oxygenated continuously in order to fuel the energy production of the cells. Oxygen in inhaled air enters the lungs, where it diffuses along the walls of the alveoli and surrounding blood capillaries and then enters the bloodstream (mainly bound to haemoglobin), which transports it to the rest of the body. This is a normal physiological process that is essential to the body’s survival.

The administration of additional oxygen in hypoxia patients will improve the supply of oxygen to the bodily tissues.

Pressurised oxygen (hyperbaric oxygen therapy) helps to significantly increase the amount of oxygen that can be absorbed into the blood (including the part not bound to haemoglobin), and, as a result, also improves the supply of oxygen to the bodily tissues.

In the treatment of gas/air embolisms, high-pressure hyperbaric oxygenation will reduce the volume of the gas bubbles. As a result, the gas can be absorbed from the bubble into the blood more effectively, and will then leave the lungs in the exhaled air.

5.2 Pharmacokinetic properties
Inhaled oxygen is absorbed in a pressure-dependent exchange of gases between the alveoli and the capillary blood that passes them.

The oxygen (mostly bonded to haemoglobin) is transported to all body tissues in the systemic circulation system. Only a very small proportion of the oxygen in the blood is freely dissolved into the plasma.

Oxygen is an essential component in the generation of energy in intermediary cell metabolism – aerobic ATP production in the mitochondria. Virtually all the oxygen absorbed by the body is exhaled as the carbon dioxide created in this intermediary mechanism.
5.3 Preclinical safety data
In animal experiments, oxidative stress has led fetal dysmorphogenesis, abortions, and intrauterine growth restriction. Excess oxygen during pregnancy may induce abnormalities in the development of the neural tube. Prolonged hyperbaric oxygen treatment during gestation in mice, rats, hamsters and rabbits was foetotoxic and teratogenic. Other animal experiments suggested that lower level exposure to hyperbaric oxygen did not have adverse developmental effects. Oxygen has shown mutagenic effects in *in vitro* tests with mammalian cells. Although available data do not suggest a tumor promoting effect for hyperbaric oxygen, conventional carcinogenicity studies are not known. As regards pharmacodynamics and toxicity after repeated administration no risks have been known to occur other than those already described in other sections.

6 PHARMACEUTICAL PARTICULARS

6.1 List of excipients
There are no excipients.

6.2 Incompatibilities
Medicinal oxygen strongly supports combustion and will cause substances to burn vigorously, including some materials that will not normally burn in air. It is highly dangerous in the presence of oils, greases, tarry substances and many plastics due to the risk of spontaneous combustion in the presence of medicinal oxygen in relatively high concentrations.

6.3 Shelf life
Gaseous medicinal oxygen may be kept up to 5 years after the date stated on the cylinder.

6.4 Special precautions for storage
- The gas cylinders should be stored between -20°C and +65°C.
- The gas cylinders should be stored vertically, except gas cylinders with a convex bottom; these should be stored horizontally, or in a crate.
- The gas cylinders should be protected from falling over or from mechanical shocks, for example, by fixing the gas cylinders or placing them in a crate.
- The gas cylinders should be stored in a well-ventilated room that is exclusively used for the storage of medicinal gases. This storage room must not contain any inflammable materials.
- Gas cylinders containing a different kind of gas, or a gas that has a different composition, should be stored separately.
- Full and empty gas cylinders should be stored separately.
- The gas cylinders must not be stored near sources of heat. If at risk of fire – move to a safe place.
- Gas cylinders must be stored covered and protected against the effects of the weather.
- Close the valves of the cylinders after use.
- Return cylinder to the supplier when empty.
- Warning notices prohibiting smoking and naked lights must be posted clearly in the storage area.
- Emergency services should be advised of the location of the cylinder storage.

### 6.5 Nature and contents of container

Gaseous medicinal oxygen is stored in gas cylinders in a gaseous state and under a pressure of 150, 200 or 300 bar (at 15°C). The cylinders are made of steel or aluminium. The valves are made of brass, steel or aluminium.

<table>
<thead>
<tr>
<th>Packaging</th>
<th>Available sizes (l)*</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aluminium cylinder with valve with integrated pressure regulation</td>
<td>1, 2, 5, 7, 10, 20, 30, 40, 47, 50</td>
</tr>
<tr>
<td>Steel cylinder with valve with integrated pressure regulation</td>
<td>1, 2, 5, 7, 10, 20, 30, 40, 47, 50</td>
</tr>
<tr>
<td>Aluminium cylinder with traditional or step down valve</td>
<td>1, 2, 5, 7, 10, 20, 30, 40, 47, 50</td>
</tr>
<tr>
<td>Steel cylinder with traditional or step down valve</td>
<td>1, 2, 5, 7, 10, 20, 30, 40, 47, 50</td>
</tr>
<tr>
<td>Steel cylinder bundles with traditional or step down valve</td>
<td>4x50, 8x50, 12x50, 16x50, 20x50</td>
</tr>
<tr>
<td>Aluminium cylinder bundles with traditional or step down valve</td>
<td>4x50, 8x50, 12x50, 16x50, 20x50</td>
</tr>
</tbody>
</table>

*7l, 40l and 47l available for 150 bar filling pressure only.

<table>
<thead>
<tr>
<th>Type of the valve</th>
<th>Outlet pressure</th>
<th>Remarks</th>
</tr>
</thead>
<tbody>
<tr>
<td>Valve with integrated pressure regulation</td>
<td>4 bar (at the socket outlet)</td>
<td></td>
</tr>
<tr>
<td>Traditional valve</td>
<td>150, 200 or 300 bar (when the gas cylinder is full)</td>
<td>Use only with a suitable reducing device.</td>
</tr>
<tr>
<td>Step down valve</td>
<td>60–70 bar</td>
<td>For 300 bar cylinders only. Use only with a suitable reducing device.</td>
</tr>
</tbody>
</table>

Gas cylinders comply with the requirements of Dir. 1999/36/EC.
Colour marking conforms to EN 1089-3: white body and white shoulder.
Valves conform to the requirements of EN ISO 10297.
Traditional and step down valves conform to NEN 3268 (NL), DIN 477 (DE), BS 341-3 (UK), NBN 226 (BE), EN ISO 407, ISO 5145.
Valves with integrated pressure regulator conform also with EN ISO 10524-3.

Gas cylinders with a content of (x) litres contain (y) kg of gas and deliver (z) m³ of oxygen at 15°C and 1 bar when filled to 150 bar.

<table>
<thead>
<tr>
<th>Content in litres (x)</th>
<th>1</th>
<th>2</th>
<th>5</th>
<th>7</th>
<th>10</th>
<th>20</th>
<th>30</th>
<th>40</th>
<th>47</th>
<th>50</th>
</tr>
</thead>
<tbody>
<tr>
<td>Content in kg (y)</td>
<td>0.217</td>
<td>0.434</td>
<td>1.086</td>
<td>1.52</td>
<td>2.17</td>
<td>4.34</td>
<td>6.51</td>
<td>8.69</td>
<td>10.21</td>
<td>10.86</td>
</tr>
<tr>
<td>Number of m³ of oxygen (z)</td>
<td>0.160</td>
<td>0.321</td>
<td>0.80</td>
<td>1.12</td>
<td>1.60</td>
<td>3.21</td>
<td>4.81</td>
<td>6.41</td>
<td>7.53</td>
<td>8.02</td>
</tr>
<tr>
<td>---------------------------</td>
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<td>------</td>
<td>------</td>
<td>------</td>
</tr>
<tr>
<td>Content in litres (x)</td>
<td>4x50</td>
<td>8x50</td>
<td>12x50</td>
<td>16x50</td>
<td>20x50</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Content in kg (y)</td>
<td>43.4</td>
<td>86.8</td>
<td>130</td>
<td>174</td>
<td>217</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Number of m³ of oxygen (z)</td>
<td>32.1</td>
<td>64.1</td>
<td>96.2</td>
<td>128.2</td>
<td>160.3</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Gas cylinders with a content of (x) litres contain (y) kg of gas and deliver (z) m³ of oxygen at 15°C and 1 bar when filled to 200 bar.

<table>
<thead>
<tr>
<th>Content in litres (x)</th>
<th>1</th>
<th>2</th>
<th>5</th>
<th>10</th>
<th>20</th>
<th>30</th>
</tr>
</thead>
<tbody>
<tr>
<td>Content in kg (y)</td>
<td>0.288</td>
<td>0.577</td>
<td>1.44</td>
<td>2.88</td>
<td>5.77</td>
<td>8.65</td>
</tr>
<tr>
<td>Number of m³ of oxygen (z)</td>
<td>0.212</td>
<td>0.425</td>
<td>1.125</td>
<td>2.12</td>
<td>4.33</td>
<td>6.37</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Content in litres (x)</th>
<th>50</th>
<th>4x50</th>
<th>8x50</th>
<th>12x50</th>
<th>16x50</th>
<th>20x50</th>
</tr>
</thead>
<tbody>
<tr>
<td>Content in kg (y)</td>
<td>14.4</td>
<td>57.7</td>
<td>115</td>
<td>173</td>
<td>231</td>
<td>288</td>
</tr>
<tr>
<td>Number of m³ of oxygen (z)</td>
<td>10.61</td>
<td>42.5</td>
<td>85.0</td>
<td>127.5</td>
<td>170.0</td>
<td>212.0</td>
</tr>
</tbody>
</table>

Gas cylinders with a content of (x) litres contain (y) kg of gas and deliver (z) m³ of oxygen at 15°C and 1 bar when filled to 300 bar.

<table>
<thead>
<tr>
<th>Content in litres (x)</th>
<th>1</th>
<th>2</th>
<th>5</th>
<th>10</th>
<th>20</th>
<th>30</th>
</tr>
</thead>
<tbody>
<tr>
<td>Content in kg (y)</td>
<td>0.413</td>
<td>0.826</td>
<td>2.06</td>
<td>4.13</td>
<td>8.26</td>
<td>12.4</td>
</tr>
<tr>
<td>Number of m³ of oxygen (z)</td>
<td>0.308</td>
<td>0.616</td>
<td>1.54</td>
<td>3.08</td>
<td>6.16</td>
<td>9.24</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Content in litres (x)</th>
<th>50</th>
<th>4x50</th>
<th>8x50</th>
<th>12x50</th>
<th>16x50</th>
<th>20x50</th>
</tr>
</thead>
<tbody>
<tr>
<td>Content in kg (y)</td>
<td>20.6</td>
<td>82.6</td>
<td>165</td>
<td>248</td>
<td>330</td>
<td>413</td>
</tr>
<tr>
<td>Number of m³ of oxygen (z)</td>
<td>15.4</td>
<td>61.6</td>
<td>123</td>
<td>185</td>
<td>246</td>
<td>308</td>
</tr>
</tbody>
</table>
Not all cylinder sizes may be marketed.

6.6 Special precautions for disposal

Preparation prior to use
Follow the instructions of your supplier, particularly:

- If the gas cylinder is visibly damaged, or if there is a suspicion of damage or exposure to extreme temperatures has occurred, the gas cylinder may not be used
- All contact with oil, grease or hydrocarbons must be avoided
- Remove the seal from the valve and the protective cap before use
- Only equipment suitable for use with a specific gas cylinder and that specific gas may be used
- Check that the quick connector and regulator are clean and that the connections are in good condition
- Open the cylinder valve slowly – at least half a turn
- When opening and closing the valve of a gas cylinder, no pliers or other tools must be used so as to avoid the risk of damage
- No modifications to the form of packaging must be made
- Check for leakage in accordance with the instructions accompanying the regulator. Do not try to deal with leakage from the valve or equipment yourself, other than by changing the gasket or O-ring
- In the event of leakage, close the valve and uncouple the regulator. If the cylinder continues to leak, empty the cylinder outdoor. Label defective cylinders, place them in an area intended for claims and return them to the supplier.
- For cylinders with an inbuilt pressure regulator valve, it is not necessary to use a separate pressure regulator. The inbuilt pressure regulator valve has a quick connector for connecting ‘on demand’ valves, but also a separate outlet for constant flow of gas, where the flow can be regulated.

Using the gas cylinder

- The transferring of gas under pressure is prohibited.
- Smoking and open flames are strictly forbidden in rooms where treatment with medicinal oxygen takes place.
- When the cylinder is in use it must be fixed in a suitable support.
- One should consider replacing the gas cylinder when the pressure in the bottle has dropped to a point where the indicator on the valve is within the yellow field.
- When a small quantity of gas is left in the gas cylinder, the cylinder valve must be closed. It is important that a small amount of pressure is left in the cylinder to avoid the entrance of contaminants.
- Valves of empty gas cylinders must be closed.
- After use the cylinder valve must be closed hand-tight. Depressurise the regulator or connection.

7 MARKETING AUTHORISATION HOLDER
SOL S.p.A.
Via Borgazzi 27
20900 Monza
Italy

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10 DATE OF REVISION OF THE TEXT
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